(CODE : 1P)



Class X Moving to Class XI (PCB)

MAT, PHYSICS, CHEMISTRY & BIOLOGY

NARAYANA SCHOLASTIC APTITUDE TEST (NSAT)

Time: 1:00 Hr.

Date : 28-11-21

Maximum marks: 140

NSAT – SET-1

IMPORTANT INSTRUCTIONS:

- 1. The test Booklet consists of 35 questions. The maximum marks are 140.
- 2. There are four parts in the question paper of MAT, Physics, Chemistry & Biology having 35 questions. Each question is allotted 4 (four) marks for each correct response.
- 3. **No Negative Marking.**
- 4. Mark only one correct answer out of four alternatives.
- 5. Use Blue/Black Ball Point Pen only for writing particulars/marking.
- 6. Use of Calculator is not allowed.
- 7. Dark the circle in the space provided only.
- 8. Use of white fluid or any other material which damage the answer sheet, is not permissible on the Answer Sheet.

TO BE FILLED IN CAPITAL LETTERS				
NAME OF THE STUDENT :				
FATHER'S NAME :				
CONTACT NUMBERS:SCHOOL NAME :				
ROLL NO. :TEST CENTRE :				
I have read all the instructions and shall abide by them	I have verified all the information filled in by the Candidate			
Signature of the Candidate	Signature of the Invigilator			



EDUCATION IS INTEGRAL FOR GROWTH AND DEVELOPMENT

Education is integral for the growth and development of an individual. The expectation from an educational institute is always about making the society better for all and to bring out one's true Potential in the service of mankind.

At Narayana, we believe that a student's education is complete only when we are able to contribute towards his/her overall development besides imparting knowledge based and career oriented training.

With an aim to provide top of the league training to students to excel in every sphere of their lives, Narayana Group has been focusing on result oriented inputs.

Narayana's courses have been designed to cater to all the needs of the aspirants to help them excel in various competitive as well as Board examinations. Innovative strategies and techniques adopted in our centres keep our students abreast of the ever-changing pattern of top level Engineering/Medical Entrance Exams. As a result, Narayana's timetested learning formulae are percolating to far-flung corners of India to benefit students from all backgrounds.

"Footprints on the sands of time are not made by sitting down". Today we rededicate the last 4 decades of our success to your dreams. I wish all our students a very successful academic year ahead.

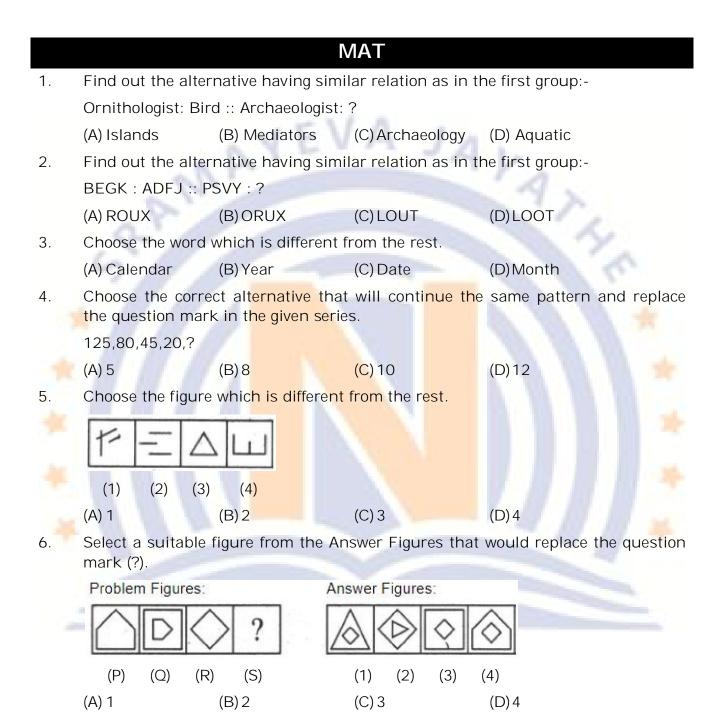
Dr. P. NARAYANA Founder, Narayana Group

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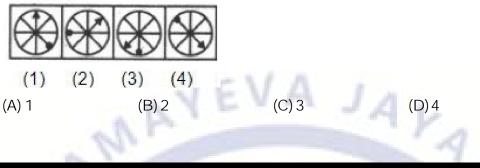
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(a:

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7. Choose the figure which is different from the rest.



- BIOLOGY
- 8. Which of the following statements is not true about compensation point?
 - (A) At compensation point, net exchange of gases is zero
 - (B) At compensation point, rate of respiration is equal to rate of photosynthesis
 - (C) Compensation point can be observed during morning as well as evening time
 - (D) Compensation point occurs when rate of transpiration is high
- 9. Which of the following statements is not true for digestion of food in human beings?
 - (A) Digestion of fats is facilitated by components present in digestive juice secreted by largest gland of the body
 - (B) Acidic chyme is made alkaline by hepatopancreatic secretions which are poured into the duodenum
 - (C) Pancreatic juice secreted from endocrine part of pancreas is a complete digestive juice as it contains enzymes for digestion of all principal dietary components
 - (D)Digestion of carbohydrates, proteins and lipids starts in oral cavity, stomach and intestine respectively and completes in small intestine

- 10. In the following questions, a statement of assertion (A) is followed by a statement of reason (R).
 - (A) If both Assertion & Reason are true and the reason is the correct explanation of the assertion, then mark (A).
 - (B) If both Assertion & Reason are true but the reason is not the correct explanation of the assertion, then mark (B).
 - (C) If Assertion is true statement but Reason is false, then mark (C).
 - (D) If both Assertion and Reason are false statements, then mark (D).
 - A : CO₂, ethyl alcohol and lactic acid are the products of anaerobic respiration.
 - R : Anaerobic respiration cannot continue for long in higher organism.
- 11. Match the following :

Column-I

- a. Amphibians
- b. Pyruvate
- c. Fish
- d. Diastolic pressure
- e. Systolic pressure
- (A) a(ii), b(iii), c(v), d(i), e(iv)
- (C) a(iii), b(ii), c(v), d(i), e(iv)

Column-II

(i) 80 mm of Hg
(ii) 3-chambered heart
(iii)3-carbon molecule
(iv)120 mm of Hg
(v) 2-chambered heart
(B)a(ii), b(iii), c(iv), d(i), e(v)
(D)a(i), b(iii), c(v), d(ii), e(iv)

- 12. Diploid cells have
 - (A) Two chromosomes

(B) One set of chromosomes

- (C) Two pairs of homologous chromosomes
- (D) Two sets of chromosomes

13.	To maintain the life of an individual organism, which of the following mentioned life processes is not necessary?	
	(A) Nutrition	(B) Respiration
	(C) Excretion	(D) Reproduction
14.	Match the following :	42
	Column-1	Column-II
	a. Paramoecium	(i) Root
	b. Plasmodium	(ii) Shoot
	c. Plumule	(iii)Multip <mark>le fissio</mark> n
	d. Radicle	(iv)Binary fission
	(A) a(iv), b(iii), c(ii),d(i)	(B)a(iii), b(iv), c(i), d(ii)
	(C)a(iv), b(iii), c(i), d(ii)	(D)a(iii), b(iv), c(ii),d(i)
15.	Match the following :	
	Column-1	Column-II
	a. Gonorrhoea, syphilis, AIDS	(i) IUD
	b. Copper-T	(ii) Saheli
	c. Non-steroidal pill	(iii)MTP
	d. Female foeticides	(iv)STDs
	(A) a(iii), b(i), c(iv), d(ii)	(B) a(iv), b(i), c(ii), d(iii)
	(C) a(iv), b(iii), c(ii), d(i)	(D) a(iv), b(i), c(iii), d(ii)

16. A man with blood group 'A' marries a woman with blood group 'AB'. Which of the following types of blood group will not be found in their offsprings?

(A) AB (B) A (C) B (D) O

17. In an experiment, Mendel bred a homozygous tall pea plant with a heterozygous tall pea plant. The plants produced in F1 generation will be

(A) 50% heterozygous tall	(B)75% homozygous tall

(C) 25% heterozygous tall (D) 75% heterozygous tall

PHYSICS					
18.	Which of the following <mark>charge</mark> is no	t possible	*		
	(A) -1.6×10^{-19} C (B) 1.6×10^{-19} C	(C) 1.6×10 ⁻²⁰ C	(D) None of these		
19.	A soap bubble is given negative charge. Its radius will				
	(A) Increase	(B) <mark>Decrease</mark>			
	(C) Remain unchanged	(D)Flu <mark>ctuate</mark>			
20.	20. Number of electrons in one micro coulomb of charge will be		vill be		
	(A) 5.46×10^{29} (B) 6.25×10^{12}	(C) 1.6×10 ¹⁹	(D) 9×10 ¹¹		

- 21. If a body is positively charged, then it has
 - (A) excess of electron (B) excess of protons
 - (C) deficiency of electron (D) deficiency of protons

- 22. Name the kind of the mirror used to obtain a virtual and diminished image
 - (A) Concave (B) Convex
 - (C) Plane (D) Both (B) and (C)

23. Two plane mirrors are inclined at an angle of 60° with each other (one is horizontal). A ray of light travelling horizontally is reflected first from one mirror and then from the other. The resultant deviation is

(A) 60° (B) 120° (C) 180° (D) 240°

24. An object is placed 40cm from a concave mirror of focal length 20cm. The image formed is

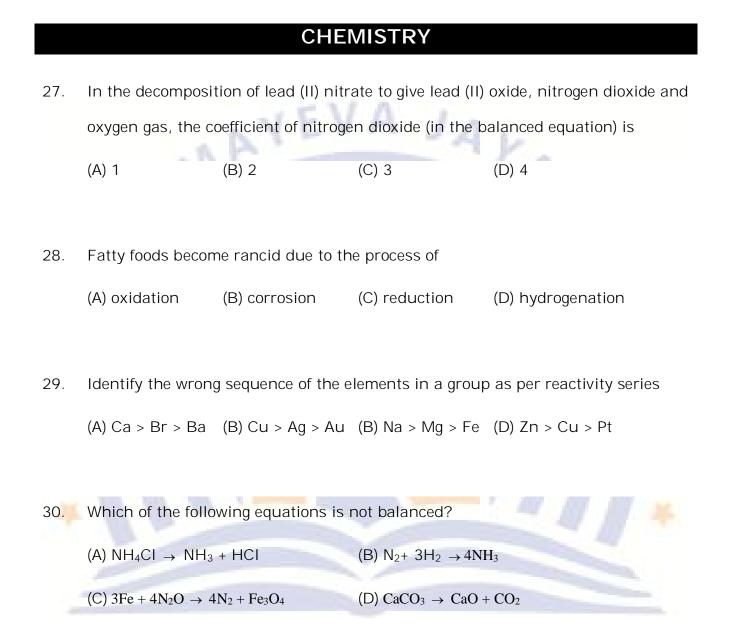
(A) Real, inverted and same in size (B) Real, inverted and smaller (C) Virtual, erect and larger (D) Virtual, erect and smaller

25. To get three images of a single object, one should have two plane mirrors at an angle of

(A) 30° (B) 60° (C) 90° (D) 150°

26. Refractive index of a substance for which critical angle in air is 45° is

(A) $\sqrt{2}$ (B) $\sqrt{3}$ (C) 1 (D) $2\sqrt{2}$



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31. Write the balance chemical equation. $FeSO_4 + H_2SO_4 + HNO_3 \rightarrow Fe_2[SO_4]_3 + NO + H_2O$

(A) 6,3,2,3,2,4 (B) 2,2,3,2 4,6 (C) 5,2,2,4,2,1 (D) 1,1,2,2,3,3

- 32. An alloy of Zn and Cu is dissolved in dil. HC1. Hydrogen gas is evolved. In this evolution of gas
 - (A) only zinc reacts with dil. HCl
 - (B) only copper reacts with dil. HCl
 - (C) both zinc and copper react with dil. HCI
 - (D) only copper reacts with water
- 33. A greenish coating develops on copper utensils due to formation of

(A) CuCO ₃	(B) Cu(OH) ₂

 $(C) Cu(OH)_2.CuCO_3 (D) CuO$

- 34. In the reaction, $CI_2 + OH^- \rightarrow CI^- + CIO_3^- + H_2O$, chlorine is :
 - (A) Oxidised (B) Reduced
 - (C) Oxidised as well as reduced (D) Neither oxidised nor reduced
- 35. There are 3 containers X, Y and Z. X contains 10ml of water and Z contains 10ml of milk. Y contains 5ml of milk (same as in container Z) mixed with 5 ml of water. All 3 containers have pH value of 6.5. P amount of Acetic acid is added to container X, Q amount to Y and R amount to Z. Such that the final pH value in each container is 5.5. Then which of the following is true

