

MAT, PHYSICS, CHEMISTRY \& BIOLOGY
NARAYANA SCHOLASTIC APTITUDE TEST (NSAT)
Time: 1:00 Hr.
Date: 28-11-21
Maximum marks: 140

## NSAT - SET-1

## IMPORTANT INSTRUCTIONS:

1. The test Booklet consists of 35 questions. The maximum marks are 140.
2. There are four parts in the question paper of MAT, Physics, Chemistry \& Biology having 35 questions. Each question is allotted 4 (four) marks for each correct response.
3. No Negative Marking.
4. Mark only one correct answer out of four alternatives.
5. Use Blue/Black Ball Point Pen only for writing particulars/marking.
6. Use of Calculator is not allowed.
7. Dark the circle in the space provided only.
8. Use of white fluid or any other material which damage the answer sheet, is not permissible on the Answer Sheet.

## TO BE FILLED IN CAPITAL LETTERS

NAME OF THE STUDENT : $\qquad$
FATHER'S NAME : $\qquad$
CONTACT NUMBERS: $\qquad$ SCHOOL NAME : $\qquad$
ROLL NO. : $\qquad$ TEST CENTRE : $\qquad$

| I have read all the instructions and shall abide by them |
| :---: |
| ............................................. |
| Signature of the Candidate |




## MAT

1. Find out the alternative having similar relation as in the first group:-

Ornithologist: Bird :: Archaeologist: ?
(A) Islands
(B) Mediators
(C) Archaeology
(D) Aquatic
2. Find out the alternative having similar relation as in the first group:BEGK : ADFJ :: PSVY : ?
(A) ROUX
(B) ORUX
(C) LOUT
(D)LOOT
3. Choose the word which is different from the rest.
(A) Calendar
(B) Year
(C) Date
(D) Month
4. Choose the correct alternative that will continue the same pattern and replace the question mark in the given series.

125,80,45,20,?
(A) 5
(B) 8
(C) 10
(D) 12
5. Choose the figure which is different from the rest.

(A) 1
(B) 2
(C) 3
(D) 4
6. Select a suitable figure from the Answer Figures that would replace the question mark (?).

Problem Figures:

(P) ( Q )

(R) (S)
(B) 2
(A) 1

Answer Figures:

7. Choose the figure which is different from the rest.

(1)
(2) (3)
(4)
(A) 1
(B) 2
(C) 3
(D) 4

## BIOLOGY

8. Which of the following statements is not true about compensation point?
(A) At compensation point, net exchange of gases is zero
(B) At compensation point, rate of respiration is equal to rate of photosynthesis
(C) Compensation point can be observed during morning as well as evening time
(D) Compensation point occurs when rate of transpiration is high
9. Which of the following statements is not true for digestion of food in human beings?
(A) Digestion of fats is facilitated by components present in digestive juice secreted by largest gland of the body
(B) Acidic chyme is made alkaline by hepatopancreatic secretions which are poured into the duodenum
(C) Pancreatic juice secreted from endocrine part of pancreas is a complete digestive juice as it contains enzymes for digestion of all principal dietary components
(D)Digestion of carbohydrates, proteins and lipids starts in oral cavity, stomach and intestine respectively and completes in small intestine
10. In the following questions, a statement of assertion (A) is followed by a statement of reason (R).
(A) If both Assertion \& Reason are true and the reason is the correct explanation of the assertion, then mark (A).
(B) If both Assertion \& Reason are true but the reason is not the correct explanation of the assertion, then mark (B).
(C) If Assertion is true statement but Reason is false, then mark (C).
(D) If both Assertion and Reason are false statements, then mark (D).

A : $\mathrm{CO}_{2}$, ethyl alcohol and lactic acid are the products of anaerobic respiration.
R : Anaerobic respiration cannot continue for long in higher organism.
11. Match the following :

## Column-I

a. Amphibians
b. Pyruvate
c. Fish
d. Diastolic pressure
e. Systolic pressure
(A) a(ii), b(iii), c(v), d(i), e(iv)
(C) a(iii), b(ii), c(v), d(i), e(iv)

## Column-II

(i) 80 mm of Hg
(ii) 3-chambered heart
(iii)3-carbon molecule
(iv) 120 mm of Hg
(v) 2-chambered heart
(B) a(ii), b(iii), c(iv), d(i), e(v)
(D) $a(i), b(i i i), c(v), d(i i), e(i v)$
12. Diploid cells have
(A) Two chromosomes
(B) One set of chromosomes
(C) Two pairs of homologous chromosomes
(D) Two sets of chromosomes
13. To maintain the life of an individual organism, which of the following mentioned life processes is not necessary?
(A) Nutrition
(B) Respiration
(C) Excretion
(D) Reproduction
14. Match the following :

Column-1
a. Paramoecium
b. Plasmodium
c. Plumule
d. Radicle
(A) a (iv), b (iii), $\mathrm{c}(\mathrm{ii}), \mathrm{d}(\mathrm{i})$
(C) a(iv), b(iii), c(i), d(ii)
15. Match the following :

Column-1
a. Gonorrhoea, syphilis, AIDS
b. Copper-T
c. Non-steroidal pill
d. Female foeticides
(A) a(iii), b(i), c(iv), d(ii)
(C) a(iv), b(iii), c(ii), d(i)
(B) $\mathrm{a}(\mathrm{iv}), \mathrm{b}(\mathrm{i}), \mathrm{c}(\mathrm{ii}), \mathrm{d}$ (iii)
(D) a(iv), b(i), c(iii), d(ii)

## Column-ll

(i) Root
(ii) Shoot
(iii)Multiple fission
(iv)Binary fission
(B) a(iii), b(iv), c(i), d(ii)
(D) a(iii), b(iv), c(ii), d(i)

## Column-ll

(i) IUD
(ii) Saheli
(iii)MTP
(iv)STDs
16. A man with blood group ' $A$ ' marries a woman with blood group ' $A B$ '. Which of the following types of blood group will not be found in their offsprings?
(A) AB
(B) A
(C) B
(D) O
17. In an experiment, Mendel bred a homozygous tall pea plant with a heterozygous tall pea plant. The plants produced in F1 generation will be
(A) 50\% heterozygous tall
(B) $75 \%$ homozygous tall
(C) $25 \%$ heterozygous tall
(D) $75 \%$ heterozygous tall

## PHYSICS

18. Which of the following charge is not possible
(A) $-1.6 \times 10^{-19} \mathrm{C}$
(B) $1.6 \times 10^{-19} \mathrm{C}$
(C) $1.6 \times 10^{-20} \mathrm{C}$
(D) None of these
19. A soap bubble is given negative charge. Its radius will
(A) Increase
(B) Decrease
(C) Remain unchanged
(D) Fluctuate
20. Number of electrons in one micro coulomb of charge will be
(A) $5.46 \times 10^{29}$
(B) $6.25 \times 10^{12}$
(C) $1.6 \times 10^{19}$
(D) $9 \times 10^{11}$
21. If a body is positively charged, then it has
(A) excess of electron
(B) excess of protons
(C) deficiency of electron
(D) deficiency of protons
22. Name the kind of the mirror used to obtain a virtual and diminished image
(A) Concave
(B) Convex
(C) Plane
(D) Both (B) and (C)
23. Two plane mirrors are inclined at an angle of $60^{\circ}$ with each other (one is horizontal). A ray of light travelling horizontally is reflected first from one mirror and then from the other. The resultant deviation is
(A) $60^{\circ}$
(B) $120^{\circ}$
(C) $180^{\circ}$
(D) $240^{\circ}$
24. An object is placed 40 cm from a concave mirror of focal length 20 cm . The image formed is
(A) Real, inverted and same in size
(B) Real, inverted and smaller
(C) Virtual, erect and larger
(D) Virtual, erect and smaller
25. To get three images of a single object, one should have two plane mirrors at an angle of
(A) $30^{\circ}$
(B) $60^{\circ}$
(C) $90^{\circ}$
(D) $150^{\circ}$
26. Refractive index of a substance for which critical angle in air is $45^{\circ}$ is
(A) $\sqrt{2}$
(B) $\sqrt{3}$
(C) 1
(D) $2 \sqrt{2}$

## CHEMISTRY

27. In the decomposition of lead (II) nitrate to give lead (II) oxide, nitrogen dioxide and oxygen gas, the coefficient of nitrogen dioxide (in the balanced equation) is
(A) 1
(B) 2
(C) 3
(D) 4
28. Fatty foods become rancid due to the process of
(A) oxidation
(B) corrosion
(C) reduction
(D) hydrogenation
29. Identify the wrong sequence of the elements in a group as per reactivity series
(A) $\mathrm{Ca}>\mathrm{Br}>\mathrm{Ba}$
(B) $\mathrm{Cu}>\mathrm{Ag}>\mathrm{Au}$
(B) $\mathrm{Na}>\mathrm{Mg}>\mathrm{Fe}$
(D) $\mathrm{Zn}>\mathrm{Cu}>\mathrm{Pt}$
30. Which of the following equations is not balanced?
(A) $\mathrm{NH}_{4} \mathrm{Cl} \rightarrow \mathrm{NH}_{3}+\mathrm{HCl}$
(B) $\mathrm{N}_{2}+3 \mathrm{H}_{2} \rightarrow 4 \mathrm{NH}_{3}$
(C) $3 \mathrm{Fe}+4 \mathrm{~N}_{2} \mathrm{O} \rightarrow 4 \mathrm{~N}_{2}+\mathrm{Fe}_{3} \mathrm{O}_{4}$
(D) $\mathrm{CaCO}_{3} \rightarrow \mathrm{CaO}+\mathrm{CO}_{2}$
31.     - Write the balance chemical equation.
$\mathrm{FeSO}_{4}+\mathrm{H}_{2} \mathrm{SO}_{4}+\mathrm{HNO}_{3} \rightarrow \mathrm{Fe}_{2}\left[\mathrm{SO}_{4}\right]_{3}+\mathrm{NO}+\mathrm{H}_{2} \mathrm{O}$
(A) 6,3,2,3,2,4
(B) 2,2,3,2 4,6
(C) 5,2,2,4,2,1
(D) 1,1,2,2,3,3
32. An alloy of Zn and Cu is dissolved in dil. HC 1 . Hydrogen gas is evolved. In this evolution of gas
(A) only zinc reacts with dil. HCl
(B) only copper reacts with dil. HCl
(C) both zinc and copper react with dil. HCl
(D) only copper reacts with water
33. A greenish coating develops on copper utensils due to formation of
(A) $\mathrm{CuCO}_{3}$
(B) $\mathrm{Cu}(\mathrm{OH})_{2}$
(C) $\mathrm{Cu}(\mathrm{OH})_{2} \cdot \mathrm{CuCO}_{3}$
(D) CuO
34. In the reaction, $\mathrm{Cl}_{2}+\mathrm{OH}^{-} \rightarrow \mathrm{Cl}^{-}+\mathrm{ClO}_{3}^{-}+\mathrm{H}_{2} \mathrm{O}$, chlorine is :
(A) Oxidised
(B) Reduced
(C) Oxidised as well as reduced
(D) Neither oxidised nor reduced
35. There are 3 containers $X$, $Y$ and $Z$. $X$ contains 10 ml of water and $Z$ contains 10 ml of milk. Y contains 5 ml of milk (same as in container $Z$ ) mixed with 5 ml of water. All 3 containers have pH value of 6.5. P amount of Acetic acid is added to container $\mathrm{X}, \mathrm{Q}$ amount to Y and R amount to Z . Such that the final pH value in each container is 5.5 . Then which of the following is true
(A) $\mathrm{P}<$ Q $<$ R
(B) $\mathrm{P}<\mathrm{R}=\mathrm{Q}$
(C) $P=Q=R$
(D) P $<$ R $<$ Q


## SENSATIONAL SUCCESS



JEE MAIN 2021
ALL INDIA OPEN CATEGORY TOP 10 RANKS


NEET- 2020


SOLID FOUNDATION GIVEN BY NARAYANA MAKES STUDENTS WORLD CHAMPIONS IN MATHS, PHYSICS, CHEMISTRY, BIOLOGY, ASTRONOMY, ENGLISH \& CYBER OLYMPIADS


